Designation: D 1133 - 09

Standard Test Method for Kauri-Butanol Value of Hydrocarbon Solvents¹

This standard is issued under the fixed designation D 1133; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon (ε) indicates an editorial change since the last revision or reapproval.

This standard has been approved for use by agencies of the Department of Defense.

1. Scope*

1.1 This test method covers the determination of the relative solvent power of hydrocarbon solvents used in paint and lacquer formulations. This test method is suitable for use with solvents having an initial boiling point over 40°C and a dry point under 300°C when determined in accordance with the procedures in Note 1.

Note 1—Test Method D 86 is used to determine the initial boiling point and dry point for mineral spirits and similar petroleum solvents. Test Method D 1078 is used for pure compounds and narrow boiling range cuts.

- 1.2 The values stated in SI units are to be regarded as standard. No other units of measurement are included in this standard.
- 1.3 For purposes of determining conformance of an observed or a calculated value using this test method to relevant specifications, test result(s) shall be rounded off "to the nearest unit" in the last right-hand digit used in expressing the specification limit, in accordance with the rounding-off method of Practice E 29.
- 1.4 For specific hazard information and guidance, consult the supplier's Material Safety Data Sheet.
- 1.5 This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.

2. Referenced Documents

2.1 ASTM Standards:²

D 86 Test Method for Distillation of Petroleum Products at Atmospheric Pressure

D 304 Specification for *n*-Butyl Alcohol (Butanol)

D 841 Specification for Nitration Grade Toluene

D 1078 Test Method for Distillation Range of Volatile Organic Liquids

E 29 Practice for Using Significant Digits in Test Data to Determine Conformance with Specifications

3. Terminology

- 3.1 Definitions:
- 3.1.1 *kauri-butanol value*, *n*—of a solvent, the volume in millilitres at 25°C of the solvent, corrected to a defined standard, required to produce a defined degree of turbidity when added to 20 g of a standard solution of kauri resin in normal butyl alcohol.
- 3.1.1.1 *Discussion*—The kauri resin solution is standardized against toluene, which has an assigned value of 105, and a mixture of 75 % *n*-heptane and 25 % toluene on a volume basis, which has an assigned value of 40.

4. Significance and Use

4.1 The kauri-butanol value is used as a measure of solvent power of hydrocarbon solvents. High kauri-butanol values indicate relatively strong solvency.

5. Apparatus

- 5.1 *Water Bath*, capable of being maintained at 25 ± 5 °C. Alternatively, a room maintained at 25 ± 5 °C may be used.
 - 5.2 Volumetric Flask, 200-mL capacity.
 - 5.3 Erlenmeyer Flask, 250-mL capacity.
- 5.4 Sample Dispensing Device, Buret, of at least 50-mL capacity, or equivalent, that is capable of accurately determining the volume dispensed to the nearest 0.1 mL.
- 5.5 *Print Specimen*—A sheet of white paper having on it black 10 or 12 point print, No. 31 Bruce old style type.

Note 2—Text in this published standard is satisfactory to use as print specimen.

6. Reagents

6.1 Purity of Reagents—Reagent grade chemicals shall be used in all tests unless otherwise specified. Unless otherwise indicated, it is intended that all reagents shall conform to the specifications of the Committee on Analytical Reagents of the

¹ This test method is under the jurisdiction of ASTM Committee D01 on Paint and Related Coatings, Materials, and Applications and is the direct responsibility of Subcommittee D01.35 on Solvents, Plasticizers, and Chemical Intermediates.

Current edition approved June 1, 2009. Published June 2009. Originally approved in 1950. Last previous edition approved in 2004 as D 1133-04.

² For referenced ASTM standards, visit the ASTM website, www.astm.org, or contact ASTM Customer Service at service@astm.org. For *Annual Book of ASTM Standards* volume information, refer to the standard's Document Summary page on the ASTM website.